

## GUAIANOLIDES FROM *ELEPHANTOPUS CAROLINIANUS*

FERDINAND BOHLMANN, NEZHUN ATEŞ (GÖREN), JASMIN JAKUPOVIC, ROBERT M. KING\* and HAROLD ROBINSON\*

Institute for Organic Chemistry, Technical University of Berlin, D-1000 Berlin 12, West Germany; \*Smithsonian Institution, Department of Botany, Washington, DC 20560, U.S.A.

(Revised received 31 August 1983)

**Key Word Index**—*Elephantopus carolinianus*; Compositae; sesquiterpene lactones; guaianolides.

**Abstract**—The roots of *Elephantopus carolinianus* afforded two guaianolides as well as two known elephantopin derivatives which were the main constituents of the aerial parts.

From *Elephantopus carolinianus* Willd., the presence of desoxyelephantopin, a type of sesquiterpene lactone which seems to be characteristic for this genus [1–8], has been reported [1]. Re-investigation of the aerial parts afforded, in addition to desoxyelephantopin [5], the C-1-epimer isodesoxyelephantopin [4] as evidenced from the  $^1\text{H}$  NMR spectrum. The roots also contained minute amounts of these lactones and two further sesquiterpene lactones, the guaianolides **1** and **2**. The structures followed from the  $^1\text{H}$  NMR spectra (Table 1) which were close to those of similar lactones [9–12]. The nature of the ester side chain at C-8 was deduced from the splitting and the chemical shifts of the corresponding signals. The stereochemistry was identical to that of 8 $\beta$ -acyloxyepreupatundin [10] and 5-desoxyeuparotin derivatives [10], respectively, as followed from the couplings observed. The presence of an  $\omega$ -coupling between H-9 and H-14 in the spectrum of **2** agreed with a 10 $\alpha$ -epoxide [12].

### EXPERIMENTAL

The air-dried plant material was extracted with Et<sub>2</sub>O–petrol (1:2) at room temp. The polar CC fractions of the extract (Et<sub>2</sub>O and Et<sub>2</sub>O–MeOH, 10:1) of the aerial parts (150 g) gave by TLC (SiO<sub>2</sub>, PF 254; detection by UV 255 nm: Et<sub>2</sub>O, two developments) 35 mg isodesoxyelephantopin and 43 mg desoxyelephantopin (increasing polarity); their  $^1\text{H}$  NMR spectra agreed with those reported previously [4, 5]. Furthermore, the structure and stereochemistry were established by spin decoupling of the 400 MHz  $^1\text{H}$  NMR spectra. The polar CC fractions (Et<sub>2</sub>O and Et<sub>2</sub>O–MeOH, 10:1) of the roots (5 g) afforded by repeated TLC (SiO<sub>2</sub> as above; Et<sub>2</sub>O–petrol, 4:1) 1 mg **1** ( $R_f$ , 0.35), 1 mg isodesoxyelephantopin ( $R_f$ , 0.5), a mixture which gave by TLC (Et<sub>2</sub>O–CH<sub>2</sub>Cl<sub>2</sub>–C<sub>6</sub>H<sub>6</sub>, 1:1:1) 1 mg desoxyelephantopin ( $R_f$ , 0.6) and 3.6 mg **2** ( $R_f$ , 0.55). Due to the minute amount, **1** and **2** could not be induced to crystallize although they were homogeneous by TLC in different solvent mixtures.

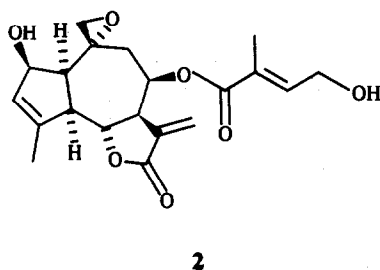
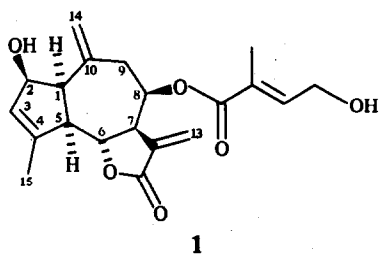
8 $\beta$ -[4'-Hydroxytigloyloxy]-preupatundin (**1**). Colourless oil, IR  $\nu_{\text{max}}^{\text{CCl}_4}$  cm<sup>-1</sup>: 3540 (OH), 1780 ( $\gamma$ -lactone), 1705 (C=CCO<sub>2</sub>R); MS  $m/z$  (rel. int.): 342 [M–H<sub>2</sub>O]<sup>+</sup> (2), 244.110 [M–RCO<sub>2</sub>H]<sup>+</sup> (6) (C<sub>15</sub>H<sub>16</sub>O<sub>3</sub>), 226 [244–H<sub>2</sub>O]<sup>+</sup> (21), 99 [RCO]<sup>+</sup> (72), 71 [99–CO]<sup>+</sup> (90), 55 (100); CI (isobutane) 361 [M+1]<sup>+</sup> (2), 245 [361–RCO<sub>2</sub>H]<sup>+</sup> (100). [ $\alpha$ ]<sub>D</sub> ~ –15, CHCl<sub>3</sub>,  $c$  = 0.1.

8 $\beta$ -[4'-Hydroxytigloyloxy]-5-desoxy-8-desacyeuparotin (**2**). Colourless oil, IR  $\nu_{\text{max}}^{\text{CCl}_4}$  cm<sup>-1</sup>: 3540 (OH), 1780 ( $\gamma$ -lactone), 1705 (C=CCO<sub>2</sub>R); MS  $m/z$  (rel. int.): 260.105 [M–RCO<sub>2</sub>H]<sup>+</sup> (3) (C<sub>15</sub>H<sub>16</sub>O<sub>4</sub>), 242 [260–H<sub>2</sub>O]<sup>+</sup> (6), 99 [RCO]<sup>+</sup> (44), 71 [99–CO]<sup>+</sup> (4), 55 (100); CI (isobutane): 377 [M+1]<sup>+</sup> (1), 261 [377–RCO<sub>2</sub>H]<sup>+</sup> (100).

Table 1.  $^1\text{H}$  NMR spectral data of compounds **1** and **2** (400 MHz, CDCl<sub>3</sub>, TMS as internal standard)

	<b>1</b>	<b>2</b>
H-1	3.20 <i>m</i>	2.06 <i>m</i>
H-2	4.73 <i>d br</i>	4.83 <i>s br</i>
H-3	5.72 <i>s br</i>	5.75 <i>s br</i>
H-5	2.67 <i>dd br</i>	2.66 <i>dd br</i>
H-6	4.65 <i>dd</i>	4.75 <i>dd</i>
H-7	3.20 <i>m</i>	3.22 <i>dddd</i>
H-8	5.53 <i>ddd</i>	5.57 <i>ddd</i>
H-9	2.86 <i>dd br</i>	2.94 <i>dd br</i>
H-9'	2.75 <i>dd br</i>	2.06 <i>m</i>
H-13	6.29 <i>d</i>	6.34 <i>d</i>
H-13'	5.50 <i>d</i>	5.55 <i>d</i>
H-14	5.12 <i>s br</i>	2.80 <i>d</i>
H-14'	5.10 <i>s br</i>	2.75 <i>dd</i>
H-15	1.99 <i>s br</i>	2.02 <i>s br</i>
OCOR	6.70 <i>tq</i>	6.71 <i>tq</i>
	4.32 <i>d br</i>	4.34 <i>d br</i>
	1.76 <i>dt</i>	1.78 <i>dt</i>

$J$  (Hz): Compound **1**: 1, 2 = 6; 1, 5 = 8; 5, 6 = 10.5; 6, 7 = 9; 7, 8 = 4; 7, 13 = 3.5; 7, 13' = 3; 8, 9 = 7; 8, 9' = 7.5; 9, 9' = 14; compound **2**: 1, 2 = 6; 1, 5 = 8.5; 5, 6 = 10.5; 6, 7 = 8.5; 7, 8 = 7, 13 = 3.5; 7, 13' = 3; 8, 9 = 8; 9, 9' = 14; 9, 14 = 1; 14, 14' = 5; OCOR: 3', 4' = 6; 3', 5' = 4'; 5' = 1.



## REFERENCES

1. Lee, K. H., Cowherd, C. M. and Wolo, M. T. (1975) *J. Pharm. Sci.* **64**, 1572.
2. McPhail, A. T. and Sim, G. A. (1972) *J. Chem. Soc. Perkin Trans. 2*, 1313.
3. Lee, K. H., Furukawa, H., Kazuka, M., Huang, H., Luhare, P. A. and McPhail, A. T. (1973) *J. Chem. Soc. Chem. Commun.* 476.
4. Govindachari, T. K., Viswanathan, N. and Führer, H. (1972) *Indian J. Chem.* **10**, 272.
5. Kuokawa, T., Nakanishi, K., Wu, W., Hsu, H., Maruyama, M. and Kupchan, S. M. (1970) *Tetrahedron Letters* 2863.
6. Kupchan, S. M., Aynechin, Y., Cassady, J. M., Schnoes, H. K. and Burlingame, A. L. (1969) *J. Org. Chem.* **34**, 3867.
7. Rustaiyan, A., Niknejad, A., Watson, W. H., Zabel, V., Mabry, T. J., Yabuta, G. and Jones, S. B., Jr. (1978) *Rev. Latinoam. Quim.* **9**, 200.
8. Lee, K. H., Ibuka, T., Furukawa, H., Kozuka, M., Wu, R. Y., Hall, I. H. and Huong, H. C. (1980) *J. Pharm. Sci.* **69**, 1050.
9. McPhail, A. T. and Sim, G. A. (1973) *Tetrahedron* **29**, 1751.
10. Bohlmann, F., Mahanta, P. K., Suwita, A., Suwita, Ant., Natsu, A. A., Zdero, C., Dörner, W., Ehlers, D. and Grenz, M. (1977) *Phytochemistry* **16**, 1973.
11. Ho, K., Sakaribara, Y. and Haruna, M. (1979) *Chem. Letters* 1473.
12. Bohlmann, F., Zdero, C., Wallmeyer, M., King, R. M. and Robinson, H., *Phytochemistry* (in press).

**Acknowledgements**—We thank Dr. E. E. Hutton of Elkins, West Virginia, for the plant material, and the Deutsche Forschungsgemeinschaft for financial support.

*Phytochemistry*, Vol. 23, No. 5, pp. 1181–1183, 1984.  
Printed in Great Britain.

0031-9422/84 \$3.00 + 0.00  
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## A GERMACRANOLIDE FROM *CYATHOCLINE LUTEA*

JAYANT S. SOHONI, BHIMSEN A. NAGASAMPAGI, JÜRGEN ZIESCHE,\* RAJINDER K. GUPTA\* and FERDINAND BOHLMANN\*

National Chemical Laboratory, Poona 8, India; \*Institute for Organic Chemistry, Technical University of Berlin, D-1000 Berlin 12, West Germany

(Revised received 31 August 1983)

**Key Word Index**—*Cyathocline lutea*; *C. lyrata*; Compositae; Astereae; sesquiterpene lactones; germacranolides; 5 $\beta$ -hydroxy-4,9-oxidogermacr-11-en-6,12-olide.

**Abstract**—While *Cyathocline lyrata* only afforded known compounds, the aerial parts of *C. lutea* gave a new sesquiterpene lactone, 5 $\beta$ -hydroxy-4,9-oxidogermacr-11-en-6,12-olide.

Two species of the small genus *Cyathocline* have been investigated chemically. While only the volatile compounds from *C. lyrata* D. Don have been studied so far [1, 2], *C. purpurea* Kuntze afforded a eudesmanolide and a guaianolide [3]. We have now studied again the aerial parts of *C. lyrata*. In addition to thymohydroquinone dimethyl ether, 10-isobutyryloxy-8,9-epoxythymol isobutyrate, parthenolide [4] and balchanin [5] were isolated.

The polar fractions of the aerial parts of *C. lutea* Cass. afforded a crystalline compound of molecular formula  $C_{15}H_{22}O_4$ ; its IR and  $^1H$  NMR spectra clearly showed the presence of a methylene lactone. Furthermore, a hydroxyl group was indicated by the IR band, as well as by the result of acetylation and oxidation, which led to the formation of a monoacetate and a ketone, respectively. Catalytic hydrogenation gave a dihydro derivative. While the signals at 60 and 90 MHz could only be partly